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#### The Mega Mole Worksheet

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Mole Calculation Worksheet - Answer Key What are the molecular weights of the following compounds? 1) NaOH  $22.99 + 16.00 + 1.01 = 40.00$  grams/mol 2) H<sub>3</sub>PO<sub>4</sub>  $3(1.01) + 30.97 + 4(16.00) = 98.00$  grams

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Related Topics: More Lessons for Chemistry Math Worksheets Mole, Mass & Avogadro Constant. An amount of substance containing  $6.02 \times 10^{23}$  particles is called a mole (often abbreviated to mol).;  $6.02 \times 10^{23}$  is called the Avogadro Constant or Avogadro's Number.; The following diagram shows how to convert between Mass, Mole and Number of particles.

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A good mixed mole problems worksheet answers will be able to show you exactly how to handle moles in various circumstances. Knowing exactly how to best proceed in any given situation is the main reason why many individuals have mixed mole problem - they lack the insight to effectively assess the situation.

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