

Fogler Solution P1

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Solutions for Chapter 1 ð Mole Balances Synopsis General: The goal of these problems are to reinforce the definitions and provide an understanding of the mole balances of the different types of reactors. It lays the foundation for step 1 of the algorithm in Chapter 4. P1-1. This problem helps the student understand the course goals and objectives.

Solutions for Chapter 1 ð Mole Balances

P1-1 Individualized solution. P1-2 (b) The negative rate of formation of a species indicates that its concentration is decreasing as the reaction proceeds ie. the species is being consumed in the course of the reaction. A positive number indicates production of the particular compound. (c) The general equation for a CSTR is: $A A A r F F V 0$

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H. Scott Fogler Ame and Catherine Vennema Professor of Chemical Engineering and The Arthur F. Thurnau Professor at the University of Michigan, ... Solutions to Chapter 1, Problems P1-1 through P1-8 Solutions to Chapter 2, Problems P2-1 through P2-10 Solutions to Chapter 3, ...

Essentials of Chemical Reaction Engineering

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P1-6 The general equation for a CSTR is: $V = FA_0 \frac{1}{1 - r_A}$. Here r_A is the rate of a first order reaction given by: $r_A = -kC_A$ Given : $C_A = 0.1C_{A0}$, $k = 0.23 \text{ min}^{-1}$, $v_0 = 10 \text{ dm}^3 \text{ min}^{-1}$, $FA = 5.0 \dots$

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Solution to Quiz #4: p1 p2 p3. Solution to Quiz #5: p1 p2. Solution to Quiz #6: p1 p2 . TextbookS (1) R equired reading: Octave Levenspiel, "Chemical Reaction Engineering," 3rd Edition, Wiley, 1999. This is one of the gems of the chemical engineering literature.

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